

Reaction of 6-Glycopyranosylaminopyrimidin-4-ones with Malonic Acids. Synthesis of 8-Glycopyranosylpyrido[2,3-d]pyrimidin-4-one Derivatives [1]

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Summary. Reactions of 6-glycopyranosylaminopyrimidin-4-ones **1 a–d** with malonic, methyl malonic and ethyl malonic acids are discussed in this paper. These reactions have been carried out in acetic anhydride at 100°C affording the 8-glycopyranosylpyrido[2,3-d]pyrimidin-4-ones **2 a–d**, **3 c–d** and the 5-acyl-6-glycopyranosylaminopyrimidin-6-ones **4 a–d**, **5 a–d**.

Keywords. Pyrido[2,3-d]pyrimidines; 8-Glycopyranosylpyrido[2,3-d]pyrimidin-4-ones; Malonic acid derivatives; Nucleosides.

Reaktion von 6-Glycopyranosylaminopyrimidin-4-onen mit Malonsäuren. Synthese von 8-Glycopyranosylpyrido[2,3-d]pyrimidin-4-on-Derivaten

Zusammenfassung. Es werden Reaktionen der 6-Glycopyranosylaminopyrimidin-4-one **1 a–d** mit Malonsäure, Methylmalonsäure und Ethylmalonsäure diskutiert. Die Reaktionen wurden in Essigsäureanhydrid bei 100°C ausgeführt und ergaben 8-Glycopyranosylpyrido[2,3-d]pyrimidin-4-one **2 a–d**, **3 c–d** und die 5-Acyl-6-glycopyranosylaminopyrimidin-6-one **4 a–d** und **5 a–d**.

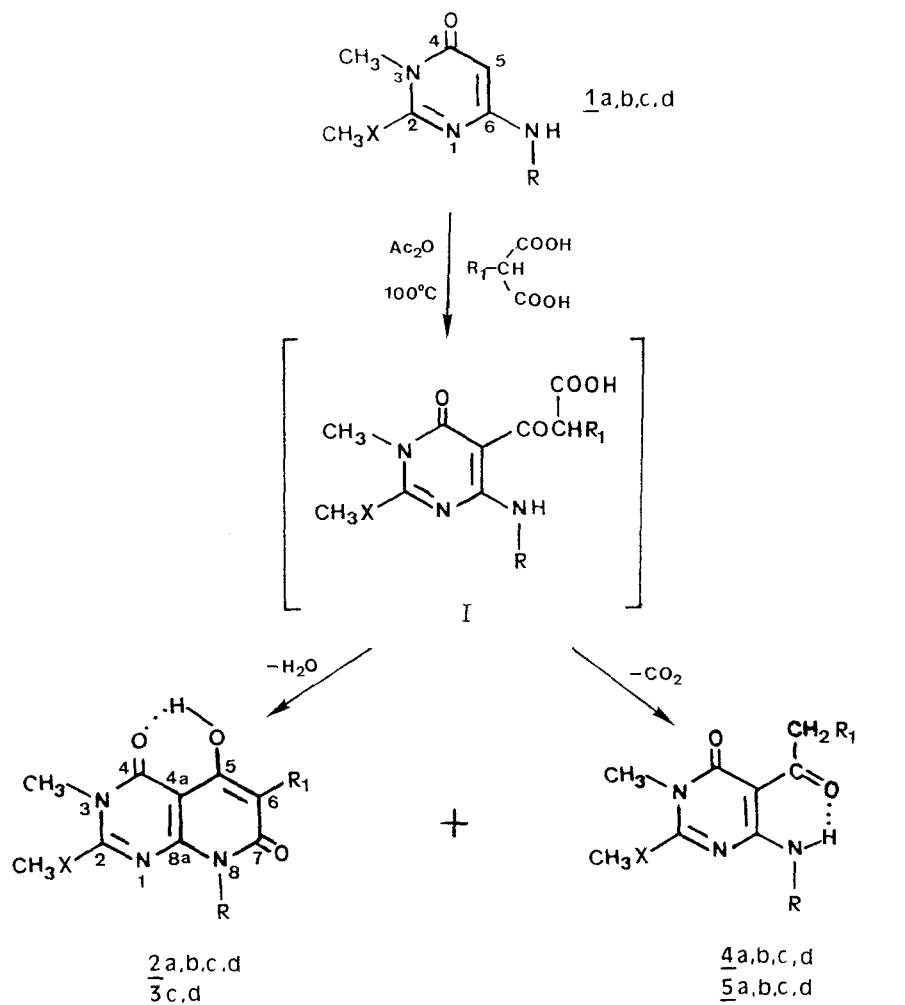
Introduction

Pyrido[2,3-d]pyrimidines are an important class of fused bicyclic derivatives known to show important biological activity [2]. One of the main synthetic ways possible for this system is the reaction of 6-aminopyrimidines with 1,3-dicarbonylic compounds [3]. This method has not been extended to 6-glycosylaminopyrimidines.

In an earlier publication, we have reported about the synthesis of some pyrano[2,3-d]pyrimidines by treatment of the corresponding 6-glycosylaminopyrimidines with malonic acid derivatives [4]. In continuation of our work on the synthesis, reactivity, and biological activity of 6-glycosylaminopyrimidines [5–9], we describe now the utility of the above mentioned method for the synthesis of several 8-glycopyranosylaminopyrido[2,3-d]pyrimidines, which can be seen as special nucleoside analogues.

Results and Discussion

The 6-glycopyranosylaminopyrimidin-4-ones **1 a-d** used in this study were obtained by a previously reported method [5]. The treatment of these compounds with methyl malonic acid and ethyl malonic acid in acetic anhydride at 100°C afforded the 8-glycopyranosylaminopyrido[2,3-d]pyrimidines **2 a-d**, **3 c-d** and the 5-acyl derivatives **4 a-d**, **5 a-b** (Scheme 1). Furthermore, in the reaction of **1 c** and **1 d** we



	X	R	R ₁
a	O	β-D-(tri-O-acetyl)xylopyranosyl	2 CH ₃
b	O	β-D-(tetra-O-acetyl)glucopyranosyl	3 CH ₂ CH ₃
c	S	β-D-(tri-O-acetyl)xylopyranosyl	4 CH ₃
d	S	β-D-(tetra-O-acetyl)glucopyranosyl	5 CH ₂ CH ₃

Scheme 1

have detected by TLC the formation of low quantities (not isolable) of 5-propionyl- (**4c-d**) and 5-butyryl-6-glycopyranosylaminopyrimidin-4-ones (**5c**). On the other hand, in the reaction of compound **1d** with ethyl malonic acid an unresolved mixture determined by $^1\text{H-NMR}$ (86% total yield) of 5-butyryl derivative **5d** (17%) and the corresponding pyrido[2,3-d]pyrimidine **3d** (69%) was obtained.

Attempts to react malonic acid were unsuccessful leading to a complex mixture with nonseparable components.

To corroborate the structures of **4a** and **4b**, direct acylations of **1a** and **1b** were carried out. When **1a** or **1b** (1 g) reacts with propionic anhydride (20 ml) in the presence of perchloric acid (1–2 drops) under reflux, 16% **4a** or 36% **4b** were obtained.

The structures **2–5** are supported by their $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ spectra. The β -configuration of the sugar moieties has been confirmed in all compounds by the values of the coupling constants $J_{1', 2'}$ and by the chemical shifts of the anomeric proton and carbon [4–10].

In the $^1\text{H-NMR}$ spectra of compounds **4** and **5** the C(6)–NH protons appear at 11.5 ppm as an exchangeable doublet. This signal appears downfield shifted in comparison with the same proton in compounds **1** because of the intramolecular hydrogen bond indicated in Scheme 1. The above signal is not observed in the spectra of compounds **2** and **3** because of the involvement of the C(6)–NH in the formation of the second heterocyclic ring. Likewise, the C(5)–OH proton signals in compounds **2** and **3** appear downfield shifted at 11.9 ppm as expected for intramolecular hydrogen bonds between this hydrogen and the C(4)=O group.

Formation of compounds **2–5** can be explained as follow. The 6-glycopyranosylaminopyrimidines **1a–d** present two active sites for electrophilic attack by malonic acids: C(5)–H and C(6)–NH. The reaction can pass through the intermediate **I** (Scheme 1) formed by attack at the C-5 position as we have observed in a similar case [10]. The β -pyrimidinyl- β -ketoacid **I** can follow two different pathways leading to the final products: formation of pyrido[2,3-d]pyrimidine derivatives **2** and **3** by a heterocyclization reaction with the C(6)–NH group and 5-acyl derivatives **4** and **5** by decarboxylation at 100°C.

We have obtained in fact both possible types of compounds: pyrido[2,3-d]pyrimidines **2** and **3** and 5-acyl derivatives **4** and **5**. In contrast to this result, pyrido[2,3-d]pyrimidines were obtained [11] when 6-amino-1,3-dimethyl uracils were acylated with malonic acid or alkyl malonic acids in the presence of acetic anhydride.

Experimental Part

Melting points (uncorrected) were determined using a melting point apparatus Gallenkamp. Proton nuclear magnetic resonance spectra were recorded with Hitachi Perkin-Elmer R-600 and Bruker AM-300 spectrometers using tetramethylsilane as internal standard. Carbon-13 nuclear resonance spectra were determined with a Bruker AM-300 spectrometer. Ultraviolet spectra were recorded with a Perkin-Elmer Lambda 5 spectrophotometer. Infrared spectra were recorded using a Beckman 4250 spectrophotometer (KBr pellets). The analysis of C, H, and N have been performed in the “Servicios Técnicos de la Universidad de Granada” using a Perkin-Elmer 240C equipment. Mass spectra were recorded using a Hewlett-Packard HP-5988-A spectrometer. Thin layer chromatography (tlc) was run on silica gel Merck 60 GF₂₅₄, visualization was accomplished by ultraviolet absorbance followed by charring with a 4% sulfuric acid-methanol solution. Finally, column chromatography was done on Kieselgel 60 silica gel (70–230 mesh).

General Procedure for the Reaction of 6-Glycopyranosylaminopyrimidines 1 a–d with Methyl and Ethyl Malonic Acids

1 g of **1 a–d** and the corresponding malonic acid derivate (double molar amount for **1 a–d** and 10 mol per mol for **1 c–d**) were added to 5 ml of acetic anhydride. The mixture was stirred at 100°C until no starting material was detected by TLC (CH₂Cl₂/MeOH, 9 : 1). After evaporation of the solvent in a rotatory evaporator the residue was dissolved in chloroform (1 ml), poured into a short chromatographic column (5 cm diameters) which contained 50 g of silica gel and eluted with CH₂Cl₂/Ethyl Ether (0–40%) in the case of **1 a, b** or CH₂Cl₂/MeOH (0–2%) in the case of **1 c, d**. The fraction containing the desired products were pooled, evaporated and crystallized from the appropriate solvent.

3,4,7,8-Tetrahydro-5-hydroxy-3,6-dimethyl-2-methoxy-4,7-dioxo-8-(2,3,4-tri-O-acetyl-β-D-xylopyranosyl)pyrido[2,3-d]pyrimidine (2 a)

Reaction time 2–3 h. Crystallized from EtOH. 0.32 g (27%), m.p. 184–186°C. ¹H-NMR (CDCl₃): δ = 1.8 [s, 3 H, C(6)–CH₃], 3.4 [s, 3 H, CH₃–N(3)], 4.0 (s, 3 H, CH₃–O), 6.6 [d, 1 H, *J* = 6.9 Hz, C(1′)–H], 11.9 [s, 1 H, C(5)–OH]. ¹³C-NMR (CDCl₃): δ = 8.6 [CH₃–C(6)], 65.6 (C-5′), 80.4 (C-1′), 92.5 (C-4 a), 103.2 (C-6), 151.8 (C-8 a), 154.7 (C-2), 161.3, 163.6, 163.9 (C-4, C-5, C-7). Anal. calc. for C₂₁H₂₅N₃O₁₁: C 50.90, H 5.08, N 8.48; found: C 50.83, H 5.16, N 8.82. [α]_D²⁵ = –70.8° (*c* = 1, CHCl₃). UV (MeOH) λ_{max} nm (ε): 234 (26 800), 290 (11 500), 308 (sh). IR (cm^{–1}): 2970 w, 1760 s, 1690 s, 1665 s, 1635 s, 1585 s, 1550 s, 1495 m, 1450 w, 1415 m, 1380 m. Mass spectrum *m/z* (abundance %): 495 (*M*⁺) (4), 373 (23), 237 (100), 236 (20), 222 (14), 97 (36), 43 (43).

3,4,7,8-Tetrahydro-5-hydroxy-3,6-dimethyl-2-methoxy-4,7-dioxo-8-(2,3,4,6-tetra-O-acetyl-β-D-glucopyranosyl)pyrido[2,3-d]pyrimidine (2 b)

Reaction time 2–3 h. Crystallized from ethyl ether. 0.35–0.47 g (30–40%), m.p. 172–174°C. ¹H-NMR (CDCl₃): δ = 1.8 [s, 3 H, C(6)–CH₃], 3.4 [s, 3 H, CH₃–N(3)], 4.0 (s, 3 H, CH₃–O), 6.6 [d, 1 H, *J* = 6.9 Hz, C(1′)–H], 11.9 [s, 1 H, C(5)–OH]. ¹³C-NMR (CDCl₃): δ = 8.6 [CH₃–C(6)], 62.0 (C-6′), 79.8 (C-1′), 92.7 (C-4 a), 103.2 (C-6), 151.8 (C-8 a), 154.7 (C-2), 161.3, 163.6, 164.0 (C-4, C-5, C-7). Anal. calc. for C₂₄H₂₉N₃O₁₃: C 50.79, H 5.15, N 7.40; found: C 51.01, H 5.20, N 7.67. [α]_D²⁵ = –39.5° (*c* = 1, CHCl₃). UV (MeOH) λ_{max} nm (ε): 223 (29 900), 290 (13 100), 311 (sh). IR (cm^{–1}): 2980 w, 2900 w, 1760 s, 1690 s, 1665 s, 1640 s, 1590 m, 1550 s, 1500 m, 1465 m, 1420 m, 1385 s. Mass spectrum *m/z* (abundance %): 567 (*M*⁺) (1), 448 (21), 348 (15), 238 (78), 237 (100), 169 (27), 109 (29), 43 (49).

3,4,7,8-Tetrahydro-5-hydroxy-3,6-dimethyl-2-methylthio-4,7-dioxo-8-(2,3,4-tri-O-acetyl-β-D-xylopyranosyl)pyrido[2,3-d]pyrimidine (2 c)

Reaction time 30 min. Crystallized from EtOH. 1.01 g (85%), m.p. 208–211°C. ¹H-NMR (CDCl₃): δ = 1.8 [s, 3 H, C(6)–CH₃], 2.8 (s, 3 H, CH₃–S), 3.5 [s, 3 H, CH₃–N(3)], 6.6 [d, 1 H, *J* = 7.4 Hz, C(1′)–H], 11.9 [s, 1 H, C(5)–OH]. ¹³C-NMR (CDCl₃): δ = 8.8 [CH₃–C(6)], 65.8 (C-5′), 80.6 (C-1′), 93.6 (C-4 a), 104.5 (C-6), 150.7 (C-8 a), 160.9 (C-2), 162.5, 163.1, 163.6 (C-4, C-5, C-7). Anal. calc. for C₂₁H₂₅N₃O₁₀S: C 49.31, H 4.93, N 8.22; found: C 49.06, H 4.96, N 8.19. [α]_D²⁰ = –63.2° (*c* = 1, CHCl₃). UV (MeOH) λ_{max} nm (ε): 227 (25 100), 295 (10 000), 333 (12 000). IR (cm^{–1}): 2970 w, 2930 w, 1755 s, 1680 s, 1650 s, 1620 m, 1530 s, 1460 m, 1415 m, 1370 m. Mass spectrum *m/z* (abundance %): 511 (*M*⁺) (9), 332 (41), 254 (91), 253 (100), 43 (78).

3,4,7,8-Tetrahydro-5-hydroxy-3,6-dimethyl-2-methylthio-4,7-dioxo-8-(2,3,4,6-tetra-O-acetyl-β-D-glucopyranosyl)pyrido[2,3-d]pyrimidine (2 d)

Reaction time 1 h. Crystallized from EtOH. 1.10 g (95%), m.p. 122–125°C. ¹H-NMR (CDCl₃): δ = 1.8 [s, 3 H, C(6)–CH₃], 2.8 (s, 3 H, CH₃–S), 3.6 [s, 3 H, CH₃–N(3)], 6.6 [d, 1 H, *J* = 7.8 Hz, C(1′)–H],

11.9 [s, 1 H, C(5)-OH]. $^{13}\text{C-NMR}(\text{CDCl}_3)$: $\delta = 8.8$ [$\text{CH}_3 - \text{C}(6)$], 61.8 (C-6'), 79.9 (C-1'), 93.7 (C-4 a), 104.5 (C-6), 150.7 (C-8 a), 161.0 (C-2), 162.5, 163.1, 163.5 (C-4, C-5, C-7). Anal. calc. for $\text{C}_{24}\text{H}_{29}\text{N}_3\text{O}_{12}\text{S}$: C 49.40, H 5.01, N 7.20; found: C 48.95, H 5.16, N 6.51. $[\alpha]_{\text{D}}^{20} = -22.1^\circ$ ($c = 1$, CHCl_3). UV (*MeOH*) λ_{max} nm (ϵ): 227 (22 000), 295 (8 700), 324–330 (10 300). IR (cm^{-1}): 2930 w, 1750 s, 1670 s, 1650 s, 1615 m, 1530 s, 1460 m, 1410 m, 1370 m. Mass spectrum m/z (abundance %): 583 (M^+) (7), 464 (46), 254 (100), 253 (76), 208 (23), 168 (26), 43 (96).

6-Ethyl-3,4,7,8-tetrahydro-5-hydroxy-3-methyl-2-methylthio-4,7-dioxo-8-(2,3,4-tri-O-acetyl- β -D-xylopyranosyl)pyrido[2,3-d]pyrimidine (3c)

Reaction time 1 h. Crystallized from *EtOH*. 0.61–0.83 g (50–68%), m.p. 175–180°C. $^1\text{H-NMR}(\text{CDCl}_3)$: $\delta = 1.1$ (t, 3 H, CH_3), 2.5 (c, 2 H, CH_2), 2.8 (s, 3 H, $\text{CH}_3 - \text{S}$), 3.6 [s, 3 H, $\text{CH}_3 - \text{N}(3)$], 6.6 [d, 1 H, $J = 7.4$ Hz, C(1')-H], 11.9 [s, 1 H, C(5)-OH]. $^{13}\text{C-NMR}(\text{CDCl}_3)$: $\delta = 12.6$ [$\text{CH}_3 - \text{CH}_2 - \text{C}(6)$], 17.0 [$\text{CH}_2 - \text{C}(6)$], 65.8 (C-5'), 80.6 (C-1'), 93.7 (C-4 a), 110.6 (C-6), 150.8 (C-8 a), 160.7 (C-2), 162.5, 163.1, 163.2 (C-4, C-5, C-7). Anal. calc. for $\text{C}_{22}\text{H}_{27}\text{N}_3\text{O}_{10}\text{S}$: C 50.28, H 5.17, N 7.99; found: C 50.10, H 5.23, N 8.20. $[\alpha]_{\text{D}}^{20} = -43.9^\circ$ ($c = 1$, CHCl_3). UV (*MeOH*) λ_{max} nm (ϵ): 228 (27 200), 295 (10 400), 323–332 (11 700). IR (cm^{-1}): 2960 w, 2930 w, 2870 w, 1760 s, 1740 s, 1675 s, 1655 s, 1615 m, 1535 s, 1460 m, 1435 m, 1410 m, 1365 m, 1325 w. Mass spectrum m/z (abundance %): 525 (M^+) (6), 406 (10), 568 (62), 267 (52), 252 (100), 43 (91).

3,4-Dihydro-3-methyl-2-methoxy-4-oxo-5-propionyl-6-(2,3,4-tri-O-acetyl- β -D-xylopyranosyl-amino)pyrimidine (4a)

Reaction time 2–3 h. Crystallized from ethyl ether. 0.18 g (16%), m.p. 190–192°C. $^1\text{H-NMR}(\text{CDCl}_3)$: $\delta = 1.1$ (t, 3 H, CH_3), 3.0 (c, 2 H, CH_2), 3.3 [s, 3 H, $\text{CH}_3 - \text{N}(3)$], 4.0 (s, 3 H, $\text{CH}_3 - \text{O}$), 5.8 [st, 1 H, + D_2O d, $J = 7.5$ Hz, C(1')-H], 11.6 [d, 1 H, $J = 8.2$ Hz, C(6)-NH]. $^{13}\text{C-NMR}(\text{CDCl}_3)$: $\delta = 8.7$ [$\text{CH}_3 - \text{CH}_2 - \text{CO} - \text{C}(5)$], 36.7 [$\text{CH}_2 - \text{CO} - \text{C}(5)$], 62.6 (C-5'), 78.5 (C-1'), 95.4 (C-5), 157.4, 162.5, 162.9 (C-2, C-4, C-6), 202.7 [$\text{CO} - \text{C}(5)$]. Anal. calc. for $\text{C}_{20}\text{H}_{27}\text{N}_3\text{O}_{10}$: C 51.16, H 5.79, N 8.95; found: C 51.03, H 5.91, N 8.94. $[\alpha]_{\text{D}}^{25} = -16.0^\circ$ ($c = 1$, CHCl_3). UV (*MeOH*) λ_{max} nm (ϵ): 226 (35 300), 261 (sh), 291 (11 400). IR (cm^{-1}): 3 200 w, 2980 w, 2930 w, 2850 w, 1745 s, 1675 s, 1635 s, 1605 m, 1555 s, 1490 m, 1455 m, 1410 w, 1380 m. Mass spectrum m/z (abundance %): 469 (M^+) (3), 350 (38), 290 (21), 240 (67), 222 (91), 210 (28), 208 (28), 194 (37), 182 (73), 97 (52), 72 (32), 43 (100).

3,4-Dihydro-3-methyl-2-methoxy-4-oxo-5-propionyl-6-(2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyl-amino)pyrimidine (4b)

Reaction time 2–3 h. Crystallized from *EtOH*. 0.40 g (36%), m.p. 158°C. $^1\text{H-NMR}(\text{CDCl}_3)$: $\delta = 1.1$ (t, 3 H, CH_3), 3.0 (c, 2 H, CH_2), 3.3 [s, 3 H, $\text{CH}_3 - \text{N}(3)$], 4.0 (s, 3 H, $\text{CH}_3 - \text{O}$), 5.7 [st, 1 H, + D_2O d, $J = 8.2$ Hz, C(1')-H], 11.4 [d, 1 H, $J = 8.2$ Hz, C(6)-NH]. $^{13}\text{C-NMR}(\text{CDCl}_3)$: $\delta = 8.8$ [$\text{CH}_3 - \text{CH}_2 - \text{CO} - \text{C}(5)$], 36.8 [$\text{CH}_2 - \text{CO} - \text{C}(5)$], 62.1 (C-6'), 79.6 (C-1'), 95.6 (C-5), 157.4, 162.4, 163.1 (C-2, C-4, C-6), 202.9 [$\text{CO} - \text{C}(5)$]. Anal. calc. for $\text{C}_{23}\text{H}_{31}\text{N}_3\text{O}_{12}$: C 51.01, H 5.77, N 7.76; found: C 50.98, H 5.57, N 7.76. $[\alpha]_{\text{D}}^{25} = +12.7^\circ$ ($c = 1$, CHCl_3). UV (*MeOH*) λ_{max} nm (ϵ): 226 (32 300), 261 (sh), 289 (10 700). IR (cm^{-1}): 3 470 w, 2970 w, 2920 w, 1745 s, 1675 s, 1615 s, 1600 s, 1555 s, 1490 m, 1450 w, 1410 w, 1370 m. Mass spectrum m/z (abundance %): 541 (M^+) (7), 422 (65), 362 (32), 348 (75), 302 (39), 210 (44), 240 (77), 222 (100), 194 (60), 182 (79), 169 (30), 141 (33), 109 (32), 43 (99).

5-Butyryl-3,4-dihydro-3-methyl-2-methoxy-4-oxo-6-(2,3,4-tri-O-acetyl- β -D-xylopyranosyl-amino)pyrimidine (5a)

Reaction time 2–3 h. Crystallized from *EtOH*. 0.47–0.76 g (40–65%), m.p. 202–204°C. $^1\text{H-NMR}(\text{CDCl}_3)$: $\delta = 1.0$ (t, 3 H, CH_3), 3.0 (c, 2 H, CH_2), 3.3 [s, 3 H, $\text{CH}_3 - \text{N}(3)$], 4.0 (s, 3 H, $\text{CH}_3 - \text{O}$), 5.8 [st, 1 H, + D_2O d, $J = 8.2$ Hz, C(1')-H], 11.6 [d, 1 H, $J = 8.9$ Hz, C(6)-NH]. $^{13}\text{C-NMR}(\text{CDCl}_3)$:

$\delta = 14.0$ [$\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CO} - \text{C}(5)$], 18.0 [$\text{CH}_2 - \text{CH}_2 - \text{CO} - \text{C}(5)$], 45.5 [$\text{CH}_2 - \text{CO} - \text{C}(5)$], 62.6 (C-5'), 78.5 (C-1'), 95.5 (C-5), 157.4 , 162.5 , 163.0 (C-2, C-4, C-6), 202.2 [$\text{CO} - \text{C}(5)$]. Anal. calc. for $\text{C}_{21}\text{H}_{29}\text{N}_3\text{O}_{10}$: C 52.16, H 6.04, N 8.69; found: C 52.21, H 5.91, N 8.97. $[\alpha]_{\text{D}}^{25} = -14.8^\circ$ ($c = 1$, CHCl_3). UV (MeOH) λ_{max} nm (ϵ): 227 (32 000), 264 (sh), 291 (11 700). IR (cm^{-1}): 2960 w, 2870 w, 1760 s, 1740 s, 1680 s, 1615 s, 1565 s, 1490 m, 1455 w, 1415 w, 1375 m. Mass spectrum m/z (abundance %): 483 (M^+) (3), 364 (42), 304 (23), 264 (52), 254 (55), 236 (89), 224 (38), 208 (49), 182 (64), 97 (54), 72 (24), 43 (100).

5-Butyryl-3,4-dihydro-3-methyl-2-methoxy-4-oxo-6-(2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyl-amino)pyrimidine (5b)

Reaction time 2–3 h. Crystallized from EtOH . 0.34 g (30%), m.p. $152\text{--}155^\circ\text{C}$. $^1\text{H-NMR}(\text{CDCl}_3)$: $\delta = 1.0$ (t, 3 H, CH_3), 3.0 (c, 2 H, CH_2), 3.3 [s, 3 H, $\text{CH}_3 - \text{N}(3)$], 4.0 (s, 3 H, $\text{CH}_3 - \text{O}$), 5.6 [st, 1 H, $+ \text{D}_2\text{O}$ d, $J = 8.9$ Hz, C(1')-H], 11.4 [d, 1 H, $J = 8.9$ Hz, C(6)-NH]. $^{13}\text{C-NMR}(\text{CDCl}_3)$: $\delta = 14.0$ [$\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CO} - \text{C}(5)$], 18.1 [$\text{CH}_2 - \text{CH}_2 - \text{CO} - \text{C}(5)$], 45.5 [$\text{CH}_2 - \text{CO} - \text{C}(5)$], 62.1 (C-6'), 79.6 (C-1'), 95.6 (C-5), 157.3 , 162.4 , 163.2 (C-2, C-4, C-6), 202.3 [$\text{CO} - \text{C}(5)$]. Anal. calc. for $\text{C}_{24}\text{H}_{33}\text{N}_3\text{O}_{12}$: C 51.88, H 5.98, N 7.56; found: C 52.19, H 5.86, N 7.85. $[\alpha]_{\text{D}}^{25} = +1.3^\circ$ ($c = 1$, CHCl_3). UV (MeOH) λ_{max} nm (ϵ): 227 (37 400), 262 (sh), 292 (12 200). IR (cm^{-1}): 2960 w, 2870 w, 1745 s, 1675 s, 1630 m, 1605 m, 1555 s, 1490 m, 1455 w, 1410 w, 1380 m, 1365 m. Mass spectrum m/z (abundance %): 555 (M^+) (1), 264 (37), 254 (22), 236 (37), 224 (31), 208 (40), 182 (39), 109 (28), 81 (36), 72 (25), 43 (100).

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